

## Review Day

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|---|-------------------|
| <b>Course Name: Physical Science, Earth Science, Biology</b>  |                   |
| <b>Unit Title: Radiation in the Human Body</b>  | <b>Day: 14/15</b> |
| <p><b>Relevant NC Standard Course of Study Goal(s):</b></p> <ul style="list-style-type: none"> <li>● <b>PSc.2.1.4 Interpret the data presented in the Bohr model diagrams and dot diagrams for atoms and ions of elements 1 through 18.</b> <ul style="list-style-type: none"> <li>○ Describe the charge, relative mass, and the location of protons, electrons, and neutrons within an atom.</li> <li>○ Calculate the number of protons, neutrons, electrons, and mass number in neutral atoms and ions.</li> <li>○ Explain how the different mass numbers of isotopes contributes to the average atomic mass for a given element (conceptual, no calculations).</li> <li>○ Explain Bohr's model of the atom.</li> </ul> </li> <li>● <b>PSc.2.3.1 Compare nuclear reactions including alpha decay, beta decay, and gamma decay; nuclear fusion and nuclear fission.</b> <ul style="list-style-type: none"> <li>○ Compare alpha, beta, and gamma decay processes –alpha decay reduces the mass of an atom by 4 and the atomic number by 2; beta decay increases the atomic number by 1 (a neutron decays into a proton and electron); gamma rays are electromagnetic waves released from the nucleus along with either an alpha or beta particle</li> <li>○ Compare the processes of fission (splitting of a very large atom) and fusion (joining of atoms) in terms of conditions required for occurrence, energy released, and the nature of products.</li> </ul> </li> <li>● <b>PSc.2.3.2 Exemplify the radioactive decay of unstable nuclei using the concept of half-life.</b> <ul style="list-style-type: none"> <li>○ Conceptually explain half-life using models</li> <li>○ Perform simple half-life calculations based on an isotope's half-life value, time of decay, and/or amount of substance.</li> </ul> </li> <li>● <b>PSc.3.1.1 Explain thermal energy and its transfer.</b> <ul style="list-style-type: none"> <li>○ Compare thermal energy, heat, and temperature.</li> <li>○ Compare conduction, convection, and radiation as methods of energy transfer.</li> </ul> </li> <li>● <b>Bio 1.1.3 Recall that chemical signals may be released by one cell to influence the development and activity of another cell.</b></li> <li>● <b>Biol 2.1.1 The input of radiant energy which is converted to chemical energy allows organisms to carry out life processes.</b> <ul style="list-style-type: none"> <li>○ Within ecosystems energy flows from the radiant energy of the sun through producers and consumers as chemical energy that is ultimately transformed into heat energy.</li> </ul> </li> <li>● <b>Bio.2.2.1 Infer how human activities (ex. pollution) may impact the environment.</b></li> <li>● <b>Bio.2.2.1 Summarize how humans modify ecosystems through population growth, technology, <b>consumption of resources and production of waste.</b></b></li> <li>● <b>Bio.3.1.3 Mutations can be random and spontaneous or caused by <b>radiation and/or chemical exposure</b></b></li> <li>● <b>EEn.1.1.3 Explain how the sun produces energy which is <b>transferred to the Earth by radiation.</b></b> <ul style="list-style-type: none"> <li>○ <b>Compare combustion and nuclear reactions (fusion and fission) on a conceptual</b></li> </ul> </li> </ul> |                   |

- level. Identify fusion as the process that produces radiant energy of stars.
- Identify the forms of energy (electromagnetic waves) produced by the sun and how some are filtered by the atmosphere (X-rays, cosmic rays, etc.).
- Summarize how energy flows from the sun to the Earth through space
- EEn.2.2.1 Explain the consequences of human activities on the lithosphere past and present.
  - Explain ways to mitigate detrimental human impacts on the lithosphere and maximize sustainable use of natural resources.
- EEn.2.2.2 Compare the various methods humans use to acquire traditional energy sources (such as peat, coal, oil, natural gas, nuclear fission, and wood).
  - Compare the methods of obtaining energy resources: harvesting (peat and wood), mining (coal and uranium/plutonium), drilling (oil and natural gas) and the effect of these activities on the environment.
- EEn.2.7.3 Explain how human activities impact the biosphere.
  - Summarize ways to mitigate human impact on the biosphere.
- EEn.2.8.1 Evaluate alternative energy technologies for use in North Carolina
  - Critique the benefits, costs and environmental impact of various alternative sources of energy for North Carolina (solar, wind, biofuels, **nuclear fusion**, fuel cells, wave power, geothermal).

### Specific Lesson Objectives

#### Students will understand:

- how energy is transferred from the Sun
- that energy can take different forms
- the scientific use of common words like heat, energy, and temperature
- how electron-electron repulsion and electron-proton attraction contributes to the Bohr theory
- how charge is neutralized in an atom
- the relative locations of parts of an atom
- how elements are organized in the periodic table
- the general process of radioactive decay on the atomic level
- how fusion fuels the Sun
- how energy is released in fusion
- the complexity of how each nation uses and harvests energy
- how energy use by nation relates to their daily lives and has differential impacts on the environment
- how different nations' energy use and North Carolina energy use is relative to their daily lives
- the complexity of energy use by region, and the complexity of attempts at mitigation of environmental damage due to harvesting of various types of energy especially at the local level
- the environmental impacts and consequences of different energy uses
- some of the evidence for the multi-faceted arguments against nuclear energy use because of disaster-driven data, and how this data is used by pro and anti-nuclear energy groups
- nuclear power is a highly efficient and “cleaner” energy source in that it does not produce carbon emissions, but there are many controversies over the long-term nuclear waste disposal from these power plants.
- the public perception of radiation on the human body and how the overall levels of radiation in everyday objects contributes to that level

- how radiation might affect the human body

**Students will know:**

- what radiation means as a scientific process
- the structure of the atom in accordance with the Bohr theory
- the components of an element symbol
- the characteristics of metals and nonmetals
- what an isotope is and how it differs from an element
- the difference between fusion and fission
- the similarities and differences between alpha, beta, and gamma decay
- Vocabulary Part I (physical science portion)
- Different energy use data by nation and how these data change by global region/country/etc.
- the types of energy and how much that energy is used in North Carolina
- the environmental costs of different types of energy harvesting processes such as drilling, fracking, nuclear power, mining for coal and uranium, etc.
- the details of the nuclear disaster of the Chernobyl nuclear power plant in Pripjat, Ukraine
- an overview of how nuclear energy use functions and how uranium is mined to fuel plants
- the definitions of nuclear waste and the nature of the radioactivity of uranium
- the standard methods for containing and managing nuclear waste in the United States, which a regional example from North Carolina
- the various radiation levels of everyday objects
- sources of radiation

**Students will be able to:**

- work in groups to answer cumulative unit questions and problems
- identify what areas they need to study independently to do well on the Unit Test

**Key Vocabulary/Formulae for this Lesson**

- all previous lesson vocabulary: energy, thermal energy, radiant energy, half-life, radiation, convection, conduction, nuclear fusion and fission, mutation, isotope, atom, Geiger counter, cancer, mutagenic, radioactivity, atomic number, proton, electron, nucleus, neutron, electromagnetic wave, chemical energy, radioactive decay, x-ray,

**Materials**

- printed study guide review packets
- stickers for work evaluation

**Technology Needs**

- none

| <b>LESSON ACTIVITIES</b>   |             |   |  |
|--|-------------|---|--|
| <b>Opening (Hook, Warm-Up, Anticipatory Set, Review, etc.)</b>   |             |   |  |
| <i>Describe activity to elicit active involvement of students or refer to previous learning:</i><br>Do Now: review questions from previous day's material (10 min) |             |   |  |
| <b>Procedure: Include all sections that apply to this lesson; combine as necessary.</b>  |             |   |  |
| <b>Section</b>   | <b>Time</b> | <b>What the Teacher will do:</b>  | <b>What the Students will do:</b>  |
| <b>Statement of Objective &amp; Purpose</b>  | 2 min       | State the goals and agenda for the day  | listen   |
| <b>Input, Modeling, &amp; Check for Understanding</b>  | 70 min      | circulate to groups and address concerns/questions from correspondents, award stickers for correct answers, | work in groups of 4 to complete study guide packet with assigned roles of 1) recorder - document group's work on packet to turn in 2) calculator - complete mathematical calculations 3) correspondent - seek help from the instructor and check completed answers 4) scout - seek help from other groups. |
| <b>Closing/ Summary</b>  | 8 min       | Give information about the test format, collect group work, and address clarifying questions on material    | listen, turn in group work, and ask questions  |
| <b>Assessment of Student Learning</b>  |             |   |  |
| <i>The Unit Test will be given next class and one packet from each group will be submitted for a participation grade.</i>  |             |   |  |